Structure and Mutagenic Conversion of E_1 Dehydrase: At the Crossroads of Dehydration, Amino Transfer, and Epimerization^{†,‡}

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ABSTRACT: Pyridoxal 5'-phosphate (PLP) and pyridoxamine 5'-phosphate (PMP) are highly versatile coenzymes whose importance is well recognized. The capability of PLP/PMP-dependent enzymes to catalyze a diverse array of chemical reactions is attributed to fine-tuning of the cofactor-substrate interactions in the active site. CDP-6-deoxy-L-threo-D-glycero-4-hexulose 3-dehydrase (E₁), along with its reductase (E₃), catalyzes the C-3 deoxygenation of CDP-4-keto-6-deoxy-D-glucose to form the dehydrated product, CDP-4-keto-3,6-dideoxy-D-glucose, in the ascarylose biosynthetic pathway. This product is the progenitor to most 3,6-dideoxyhexoses, which are the major antigenic determinants of many Gram-negative pathogens. The dimeric [2Fe-2S] protein, E₁, cloned from Yersinia pseudotuberculosis, is the only known enzyme whose catalysis involves the direct participation of PMP in one-electron redox chemistry. E_1 also contains an unusual [2Fe-2S] cluster with a previously unknown binding motif (C-X₅₇-C-X₁-C-X₇-C). Herein we report the first X-ray crystal structure of E₁, which exhibits an aspartate aminotransferase (AAT) fold. A comparison of the E₁ active site architecture with homologous structures uncovers residues critical for the dehydration versus transamination activity. Site-directed mutagenesis of four E₁ residues, D194H, Y217H, H220K, and F345H, converted E₁ from a PMP-dependent dehydrase to a PLP/glutamatedependent aminotransferase. The E₁ quadruple mutant, having been conferred this altered enzyme activity, can transaminate the natural substrate to CDP-4,6-dideoxy-4-amino-D-galactose without E₃. Taken together, these results provide the molecular basis of the functional switch of E₁ toward dehydration, epimerization, and transamination. The insights gained from these studies can be used for the development of inhibitors of disease-relevant PLP/PMP-dependent enzymes.

Vitamin B₆-derived pyridoxal 5'-phosphate (PLP)¹ and pyridoxamine 5'-phosphate (PMP) are coenzymes with extraordinary catalytic versatility (*I*). PLP/PMP-containing enzymes catalyze diverse chemical reactions such as transamination, dehydration, racemization, decarboxylation, and β , γ -elimination/substitution. The catalytic versatility of this class of enzymes is largely controlled by the specific coenzyme—substrate interactions in the active site, where

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the chemical properties of the coenzyme, modulated by the surrounding residues, determine the outcomes of the reactions (2). Because PLP/PMP-containing enzymes often play critical roles in pathogenesis and metabolism, they have been targeted for potential therapeutics to treat conditions such as cancer (3), infectious diseases (4), and Parkinson's disease (5). Because the success of drug development commonly relies on exploiting the catalytic properties of the target enzymes, such that the drug inhibits only the intended target, understanding the molecular basis of reactions catalyzed by PLP/PMP-dependent enzymes not only is of scientific interest but may also produce new therapeutic agents.

The 3,6-dideoxyhexoses, found predominantly in the *O*-antigen of lipopolysaccharides (LPS) (6), have been shown to be the major antigenic determinants of the Gram-negative bacteria (7). In nature, CDP-6-deoxy-L-threo-D-glycero-4-hexulose-3-dehydrase (E₁), along with its reductase (E₃), catalyzes the C-3 deoxygenation of CDP-4-keto-6-deoxy-D-glucose (1) in the presence of NADH to form CDP-4-keto-3,6-dideoxy-D-glucose (2) in the 3,6-dideoxyhexose biosynthetic pathway (Figures 1 and 2) (8). Product 2 is the common precursor to at least four of the seven naturally occurring 3,6-dideoxyhexoses, namely, abequose, ascarylose, paratose, and tyvelose (Figure 2) (8). The importance of 3,6-dideoxyhexoses to Gram-negative pathogenicity makes E₁

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[‡] Coordinates and observed structure factor amplitudes have been deposited in the Protein Data Bank (PDB codes 3BCX and 3BB8).

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¹ Abbreviations: PLP, pyridoxal 5'-phosphate; PMP, pyridoxamine 5'-phosphate; AAT, aspartate aminotransferase; LPS, lipopolysaccharide; NADH, nicotinamide adenine dinucleotide; rmsd, root mean squared deviation; CDP, cytidine diphosphate; ESI-MS, electrospray ionization mass spectrometry; IPTG, isopropyl β -D-thiogalactoside, DTT, dithiothreitol; LB, Luria–Burtani; DEAE, diethylaminoethyl; SDS–PAGE, sodium dodecyl sulfate–polyacrylamide gel electrophoresis; HPLC, high-performance liquid chromatography; α-KG, α-ketoglutarate; PDB, Protein Data Bank.

FIGURE 1: Starting with CDP-4-keto-6-deoxy-D-glucose (1), the E_1 dehydrase catalyzes a PMP-dependent dehydration initiated by the formation of the external aldimine 3, followed by β -elimination of the 3-OH group to give the aldimine intermediate 4 and then reduction by both E_3 and E_1 involving electron relay through FAD and [2Fe-2S] clusters to produce the product CDP-4-keto-3,6-dideoxy-D-glucose (2). This product is a common intermediate in the biosynthesis of 3,6-dideoxy sugars, such as abequose, paratose, ascarylose, and tyvelose, that are important for Gram-negative bacterium pathogenesis.

an appealing target for inhibitors that could attenutate the effects of this Gram-negative bacterial arsenal. Crystal structures of E_1 will facilitate the development of E_1 -inhibiting molecules.

 E_1 is homologous to proteins in the PLP-containing aspartate aminotransferase (AAT) superfamily (9). Many of them serve as dehydrases or aminotransferases in nucleotide sugar biosynthetic pathways (10–13). Interestingly, E_1 contains a [2Fe-2S] cluster in addition to PMP (14), distinguishing E_1 from these other enzymes. This unusual structural feature is even more intriguing, because the cysteine ligands of the iron—sulfur cluster have a C-X₅₇-C-X₁-C-X₇-C motif, which is unique among [2Fe-2S]-containing enzymes (15). Determination of the crystal structure of E_1 can help to verify the configuration of this unique [2Fe-2S] cluster and provide a structural basis to elucidate the PMP-radical mechanism distinctly observed in E_1 .

In this work, we present the first crystal structures of the wild-type E_1 , the PLP-bound H220K E_1 mutant protein, and a sugar substrate-bound H220K E_1 . Previously, H220 was proposed to be an important catalytic residue (I6). Because of nearly invariable sequence overlap of a lysine in this position across homologous aminotransferases, the H220K mutation was generated to determine if this singular alteration could convert E_1 from a dehydrase to an aminotransferase (I7). Indeed, the H220K mutation changed E_1 from a PMP-dependent dehydrase to a PLP-dependent, noncatalytic aminotransferase (I7). However, the molecular nature that

distinguishes a dehydrase versus an aminotransferase is not clear, and additional E₁ residues need to be mutated to generate a catalytic aminotransferase. The wild-type and H220K E₁ crystal structures provide molecular resolution of the catalytic residues involved and shed light on factors governing substrate specificity. Moreover, the crystal structures identify a histidine (H278) instead of a cysteine (C192 or C193) residue as the fourth ligand of the unique [2Fe-2S] motif in E₁. This information offers considerable insight into the catalytic mechanism of E₁ and enables a critical evaluation of the differences between a dehydrase and a transaminase in vitamin B₆-dependent enzymes. On the basis of these structures, mutagenesis of four active site residues successfully transformed E₁ from a PMP-mediated C-3 dehydrase into a PLP/L-glutamate-dependent C4 aminotransferase (Figure 2). Together, these results provide a structural rationale for the chemistry carried out by the E_1-E_3 complex and lay the foundation for future development of inhibitors able to attenuate infectivity of Gram-negative pathogens.

MATERIALS AND METHODS

Chemicals and DNA Manipulation. All chemicals were purchased from Sigma-Aldrich Chemical Co. and were of the highest grade available. The H220K mutation (pTrc99A) was generated previously (17). The four mutations were constructed in three PCR rounds by using the Quik-Change site-directed mutagenesis kit (Stratagene). Synthetic oligo-

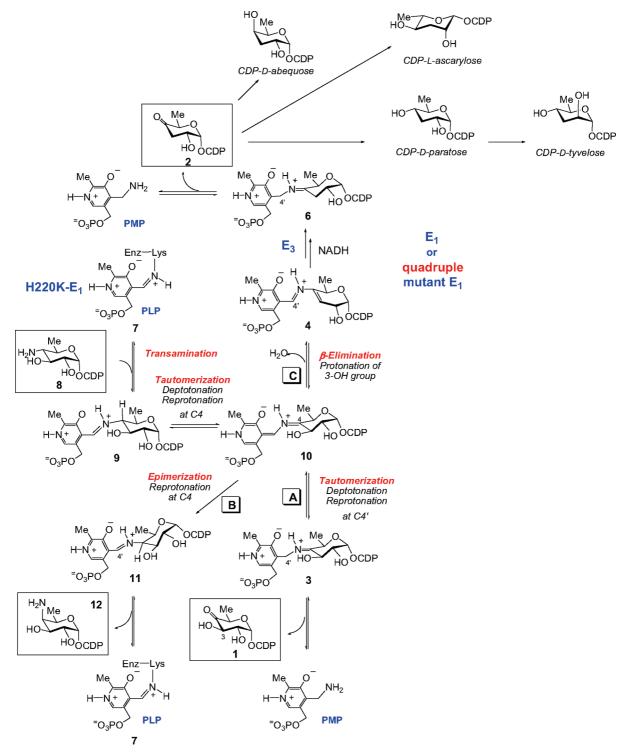


FIGURE 2: The versatility of reactions catalyzed by E1 and its mutants, in addition to representative 3,6-dideoxyhexoses produced from subsequent biosynthetic steps.

nucleotides were from Operon. The following primers, from 5' to 3', were used for mutagenesis: D194H, GAAGACT-GCTGCCATGCGTTGGGTTCC; Y217H, GGTACCGT-TAGCTTCCATCCCGCTAAACATATC; F345H, GATC-CATCATGGCATGGTTTCCCTATCACTCTG. Reverse primers were inverse complements of the forward primers. Point mutations were confirmed by sequence analysis. Strain BL21 λ (DE3) was used for recombinant protein expression.

Protein Expression and Purification. The wild type and H220K mutant of E₁ follow the same procedure as described previously (17, 18). Unlike the wild type and H220K variants of E_1 , the E_1 quadruple mutant purification was carried out aerobically on the basis of the proposed mechanism where the iron-sulfur cluster was not of concern for the speculated chemistry, though 5 mM DTT was included in all buffers. The recombinant E₁ quadruple mutant was expressed in Escherichia coli. Cultures were grown in LB broth with 100 mg/L ampicillin. At $OD_{600} = 0.6-0.8$, protein expression was induced by 0.1 mM isopropyl β -D-thiogalactoside (IPTG), and the cultures were grown at 18 °C overnight. Cells were harvested by centrifugation (5000g for 20 min), followed by resuspension, sonication, and centrifugation for

Table 1: Crystallization, Data Collection, and Refinement Statistics of E1

	E ₁ -wild type	E ₁ -H220K	E ₁ -H220K-substrate	
	(A) (Crystallization		
	0.1 M Hepes, 2% (v/v) PEG 400, 2 M ammonium sulfate, 2% (w/v) benzamidine hydrochloride (pH 7.5)	0.1 M Hepes, 2% (v/v) PEG 400, 2 M ammonium sulfate, 2% (w/v) benzamidine hydrochloride (pH 7.5)	0.1 M Hepes, 2% (v/v) PEG 400, 2 M ammonium sulfate, 2% (w/v) benzamidine hydrochloride (pH 7.5)	
	(B) Crys	stallographic Data		
space group cell dimensions	P3 ₂	P3 ₂	P3 ₂	
a, b, c (Å) $\alpha = \beta, \gamma$ (deg) resolution (Å) mosaicity (deg)	97.31, 97.31, 142.35 90, 120 50.0-2.40 0.30	98.18, 98.18, 140.39 90, 120 50.0-2.35 0.28	98.43, 98.43, 139.77 90, 120 50.0-3.05	
no. of observations no. of unique reflections completeness (%) (last shell)	435918 60161 100 (100.0)	0.28 2069783 63011 99.9 (100.0)	890792 28217 97.3 (98.1)	
$I/\sigma(I)$ (last shell) R_{merge} (%) (last shell)	15 (3.9) 11.6 (42.8)	28.3 (3.3) 4.5 (44.7)	19.8 (8.2) 7.2 (20.1)	
	(C)	Refinement		
resolution (Å) no. of reflections no. of protein atoms no. of cofactor atoms no. of ligand atoms no. of waters $R_{\rm free}$ (%) $R_{\rm crys}$ (%)	50.0-2.40 55314 3301 1 0 200 23.1 20.8	50.0-2.35 58574 3265 15 0 241 25.0 21.9	50.0-3.05 26399 3265 15 (hypothetically) 32 (hypothetically) 0 25.9 21.5	
	(D) Geometry		
rms bonds (Å) rms angles (deg) rms B main chain rms B side chain	0.007 1.32 1.42 2.15	0.007 1.30 1.39 1.89	0.008 1.37 1.30 2.94	
Ramachandran plot (%) most favored favored generously allowed	86.8 12.1 0.6	86.9 12.1 0.4	81.5 17.3 0.7	

cell debris removal (18000g for 1 h). The lysis buffer was 50 mM Tris-HCl, pH 7.5 (buffer A). The solution was loaded onto DEAE-Sepharose media preequilibrated with buffer A. After being washed with buffer A, a linear gradient of NaCl from 0 to 500 mM in buffer A was run with a gradient maker via gravity. Brown fractions were collected and dialyzed in 4 L of buffer A overnight. The sample was concentrated and passed over a Mono Q 5/50 GL column. A linear gradient from 0 to 100% buffer B (buffer A made 0.5 M in NaCl) was run over 20 column volumes. The A_{280} displayed multiple overlaid peaks in the region of the suspected protein. SDS-PAGE confirmed the inclusion of multiple proteins in the sample.

Crystallization and Data Collection. Crystals for the wildtype and H220K mutant E₁ were grown in anaerobic and aerobic environments, respectively, at 2 mg/mL in 0.1 M Hepes (pH 7.6), 1.0 M ammonium sulfate, 2% (v/v) PEG 400, and 2% (v/v) benzamidine HCl by the sitting drop vapor diffusion method at 25 °C. Substrate-bound crystals of the H220K mutant E₁ were grown as the H220K mutant E₁ with the exception that the protein sample was mixed with 1 mM (50-fold molar excess) synthetically made CDP-4,6-dideoxy-4-aminoglucose (9) (7) and left to incubate on ice for 1 h prior to adding the sample to the crystal tray. Prior to diffraction, crystals were cryoprotected in 2 M lithium sulfate and frozen in liquid N₂. Diffraction data were collected at the Stanford Synchrotron Radiation Laboratory (SSRL) and Advanced Light Source (ALS) and integrated and reduced using HKL2000 (20) (Table 1).

Molecular Replacement and Refinement. The wild-type E_1 crystal structure was determined by molecular replacement in CNS, using the crystal structure of AHBA synthase as the search model (13, 21). After rebuilding with Quanta, further refinement was performed in CNS using torsion angle simulated annealing followed by energy minimization and positional and individual B-factor refinement. Subsequent rounds of model building and refinement were carried out with the maximum likelihood approach implemented within CNS. Water molecules were added and edited both visually and with an automated water picking program (CNS) to afford an R_{crys} of 20.8% ($R_{\text{free}} = 23.1\%$) (Table 1) (21). The $H220K\ E_1$ and the substrate-bound $H220K\ E_1$ structures were solved via molecular replacement using the wild-type structure as the search model and refined by CNS to afford an R_{crys} of 21.9% ($R_{\text{free}} = 25.0\%$) and 21.5% ($R_{\text{free}} = 25.9\%$), respectively (Table 1).

Location of the Singular Iron Site on the Wild-Type E_1 Structure. Upon Synchrotron irradiation of the crystal (backsoaked in iron-free cryoprotectant prior to freezing) at the iron-absorption edge, a strong fluorescence signal was detected, indicating the presence of iron in the crystal. However, the disordered nature of the iron-binding E_1 loop precludes the precise location of the iron site. However, large anomalous peaks were observed from the anomalous difference map [FT (fom ($|f_a+|-|f_a-|$) exp(i[phase-90]))] calculated by CNS, using the experimentally determined peak wavelength (1.73970 Å) of the three-wavelength data set (Table S1, Supporting Information). The phase includes the

model phase calculated from refined E₁ model. A simulated annealing composite omit map was subsequently generated by CNS that corroborated the peak from the anomalous FT map with a defined patch of electron density outside the current protein model and overlaying perfectly on the highest anomalous difference peak (5.62 I/σ) that is located near the end of the disordered loop, where iron is expected to

Quantitation of Enzyme-Bound Iron in Wild-Type E_1 and the H278A and H278C Mutants. The enzyme-bound iron was determined by a literature procedure (19). The standard curve was determined using $Fe(NH_4)_2(SO_4)_2$. Protein samples (1 mL each) for iron titration analysis were mixed with 500 μL of reagent A (1:1 of 4.5% KMnO₄:1.2 N HCl) and incubated at 60 °C for 2 h. To these samples was added 100 μL of reagent B (8.8 g of ascorbic acid, 9.7 g of ammonium acetate, 80 mg of ferrozine, 80 mg of neocuproine, and ddH₂O to 25 mL total volume) followed by immediate vortexing. The absorbance of the samples was determined at 562 nm after the samples had been incubated for 1 h at room temperature. Purified E₁ contained approximately 1.95 iron per monomer, consistent with E₁ containing a [2Fe-2S].

Activity Assay of the H278A and H278C Mutants. The activity of wild-type, H278A, and H278C E₁ was determined by a previously published procedure (16). Briefly, the activity was determined by an assay coupling E₁ with E₃. The assay mixture consists of 25 μ M PMP, 200 μ M NADH, 100 μ M CDP-4-keto-6-deoxy-D-glucose (1), and an appropriate amount of E_3 in 800 μ L of 50 mM potassium phosphate buffer (pH 7.5). The reaction was initiated by the addition of E_1 , and the E₁ activity was determined by measuring the rate of decrease of the absorbance at 340 nm ($\epsilon_{340} = 6220 \text{ M}^{-1}$ cm⁻¹) within the initial 1 min. Each reading was calibrated against background activity (about 2% corresponding to the E₃-catalyzed oxidation of NADH in the presence of O₂) which was recorded prior to the addition of E₁. The E₁-induced NADH oxidation which was measured individually was also subtracted from the observed readings. The specific activity of wild-type E₁ was 85 units/mg of protein, which was comparable to those previously reported.

PLP Reconstitution of the E_1 Quadruple Mutant. The purified E₁ quadruple mutant was reconstituted with a 10fold excess of PLP at 4 °C for 1 h and subsequently dialyzed against 50 mM potassium phosphate buffer (pH 7.5) with four times of buffer change. The reconstituted protein was concentrated by ultrafiltration through YM10 Diaflo membrane (Amicon).

Activity Assay of the E_1 Quadruple Mutant. Activity assays were carried out in 50 mM potassium phosphate buffer, pH 7.5, in the presence of 1 mM substrate (CDP-4-keto-6-deoxy-D-glucose, 1), 50 μ M E₁ quadruple mutant, 1.5 mM PLP, and 4.5 mM L-glutamate. The final volume of the assay mixture was 0.5 mL. The reaction mixture was incubated at 25 °C, and the reaction was then stopped at different time points by removing the enzymes by ultrafiltration through a YM10 Diaflo membrane (Amicon). The reaction progress was monitored by HPLC on a CarboPac PA1 anion-exchange column (4 \times 250 mm) obtained from Dionex (Sunnyvale, CA). The flow rate was 1.0 mL/min, and the detector was set at 267 nm. A linear gradient from 2.5% to 10% buffer B (1 M ammonium acetate, pH 7.0) in buffer A (ddH₂O) over 20 min, followed by a second linear gradient from 10% to

Table 2: Activity of the E₁ Quadruple Mutant^a

	E-PLP	E-PLP $(A_{430})^b$		E-PMP $(A_{340})^b$		E-PLP (A ₄₃₀) ^c	
parameter	value	std error	value	std error	value	std error	
A	-0.3009	0.0093	0.3083	0.0091	0.2073	0.0034	
$k_{\rm min}^{-1}$	0.0277	0.0021	0.0266	0.0020	0.0035	0.0001	
C	0.6818	0.0084	0.3635	0.0082	0.3523	0.0033	

^a Data fit with the equation $y = A(1 - e^{-kt}) + C$, where y = the observed signal, k = the apparent first-order rate constant, and t = time. ^b First half-reaction. ^c Second half-reaction.

30% buffer B in buffer A over 20 min, gave a satisfactory separation between substrate 1 and the new product (at around the retention time 10 min). The new product was collected and subjected to mass spectrometry analysis. The new product was identified as CDP-4,6-dideoxy-4-aminogalactose (12) on the basis of its retention time, which was identical to that of a chemically synthesized standard (17), as well as high-resolution ESI-MS data (calcd for $C_{15}H_{25}N_4O_{14}P_2$ [M·H]⁻, 547.0843; found, 547.0841). The maximum percent conversion was 32% after incubation at 25 °C for 27 h. The small bump eluted at around 25 min in the HPLC trace is a decomposition product, CMP, derived from the substrate during HPLC analysis.

Steady-State Kinetics of the E_1 Quadruple Mutant. (A) The First Half-Reaction. The E₁ quadruple mutant was reconstituted with a 10-fold excess of PLP by incubating the enzyme and PLP at 4 °C for 1 h. Subsequently, the excess amount of PLP was removed by dialysis against 50 mM KP_i buffer (pH 7.5) containing 1 mM DTT with three times of buffer change. The reaction mixture contained 25 μ M reconstituted enzyme and 20 mM L-Glu in 50 mM KP_i buffer (pH 7.5) containing 1 mM DTT. The total reaction volume was 250 μ L, and the reaction mixture was incubated at 25 °C. Reaction progress was followed by monitoring the absorbance maxima of the PMP and PLP cofactor forms at 340 and 430 nm, respectively. Resulting time courses were fit to the equation $y = A(1 - e^{-kt}) + C$, where y = the observed signal, k = the apparent first-order rate constant, and t = time.

(B) The Second Half-Reaction. The E_1 quadruple mutant was reconstituted with a 10-fold excess of PMP by incubating the enzyme and PMP at 4 °C for 1 h. Subsequently, the excess amount of PMP was removed by dialysis against 50 mM KP_i buffer (pH 7.5) containing 1 mM DTT with three times of buffer change. The reaction mixture contained 25 μM reconstituted enzyme and 270 μM sugar substrate 1 in 50 mM KP_i buffer (pH 7.5) containing 1 mM DTT. The total reaction volume was 250 μ L, and the reaction mixture was incubated at 25 °C. Reaction progress was followed by monitoring the increase of the absorbance at 430 nm (the PLP cofactor form). Resulting time courses were fit to the equation $y = A(1 - e^{-kt}) + C$, where y = the observed signal, k = the apparent first-order rate constant, and t = time.

The activity data are summarized in Table 2.

RESULTS AND DISCUSSION

Overall Fold. Similar to the seven homologous enzymes (13, 22–27), E₁ and the H220K mutant have folds typical of the AAT superfamily. When the Fe-S-coordinating loop is excluded, the wild-type and H220K E₁ structures can be superimposed with an rmsd of 1.23–1.4 Å. However, the

FIGURE 3: (A) Overall topology of dimeric H220K E₁, including the cofactor binding site and the putative [2Fe-2S]-binding loop. (B) One H220K E₁ monomer labeled with all secondary structure elements. The cofactor PLP is shown in spheres.

wild-type structure does not contain the cofactor, while the cofactor PLP is clearly present in the H220K structure. For this reason, we display the H220K structure in Figure 3 instead of the wild-type structure (2). Both the wild-type and the H220K E₁ exist as dimers in the crystal form, similar to results obtained in solution and reported previously (Figure 3A). In both structures, the two monomers are nearly identical, with an rmsd <0.25 Å. E₁ has a conserved N-terminal large cofactor binding domain and a C-terminal small domain (Figure 3B). The N-terminal helices (A, B, C) precede the large domain, which has a central sevenstranded β -sheet (strand order 1-7-6-5-4-2-3) with strand 7 antiparallel to the remaining strands. The central β -sheet is surrounded by three α -helices (C, G, F) opposite to the active site opening, along with four α-helices (H, D, E, K) facing the active site pocket opening; the macrodipole of helix D may play a role in stabilizing the phosphate moiety of the cofactor. After the large cofactor domain is a Fe-S-coordinating loop (residues 249–280, with 253–268 disordered), which then leads into the C-terminal domain consisting of four helices (I, J, K, L) and strands 8-10. Clearly, E₁ has all of the hallmark structural features of the AAT superfamily, though it distinguishes itself from this enzyme family with the presence of the unique Fe-S-coordinating loop (10, 11, 28).

Cofactor Binding Site. The wild-type active pocket, though lacking the PMP cofactor, exhibits a PMP-binding architecture similar to PMP-bound homologous structures, such as BtrR (23), ArnB (25), and PseC (27). The residues defining the PLP/PMP-binding motif, including S86, G87, S88, F120, T122—123, D191, S215, E226, G227, W247, R264, N288, and F377, are highly conserved (Figure 5). These residues in the apo wild-type E₁ and PLP-H220K E₁ structures are almost superimposable (Figure 4A). The PLP-H220K E₁ structure shows well-defined PLP electron density as a covalent Schiff base adduct between the amino group of K220 and the C4' carbonyl group of PLP (internal aldimine, Figure 4B). The lysine aldimine group is 35 deg out of plane of the PLP pyridine ring and 2.75 Å away from the 3'-oxygen

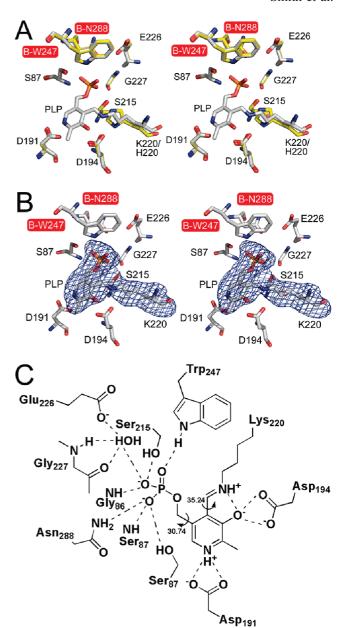


FIGURE 4: (A) Overlay of wild-type and H220K residues in the E_1 active site showing the highly similar architecture. (B) Well-defined PLP/internal lysine aldimine electron density ($F_0 - F_c$ omit map, contoured at 3σ). (C) Scheme of residues in the wild-type E_1 active site.

of the ring (salt bridge) (Figure 4C). This PLP—enzyme binding motif is consistent with homologous PLP-containing enzyme structures (13, 22–27). Therefore, although the wild-type structure of E_1 was solved without well-defined electron density of the natural cofactor PMP, the PMP docking position in the wild-type enzyme is expected to mirror the PLP position in H220K E_1 .

Similar to homologous enzyme structures, the E₁ active site is composed of residues from both monomers (Figure 4) (9). In the PMP/PLP-binding pocket, the conserved W247 from one monomer aids in stabilizing the PLP phosphate of the opposite monomer via side chain hydrogen bonding. In addition, the G86 backbone nitrogen, S87 side chain and backbone nitrogen, and the S215 side chain all directly hydrogen bond with the PMP/PLP phosphate, while the E226 side chain and G227 backbone indirectly help to stabilize the same phosphate via an ordered water (Figure 4C). The

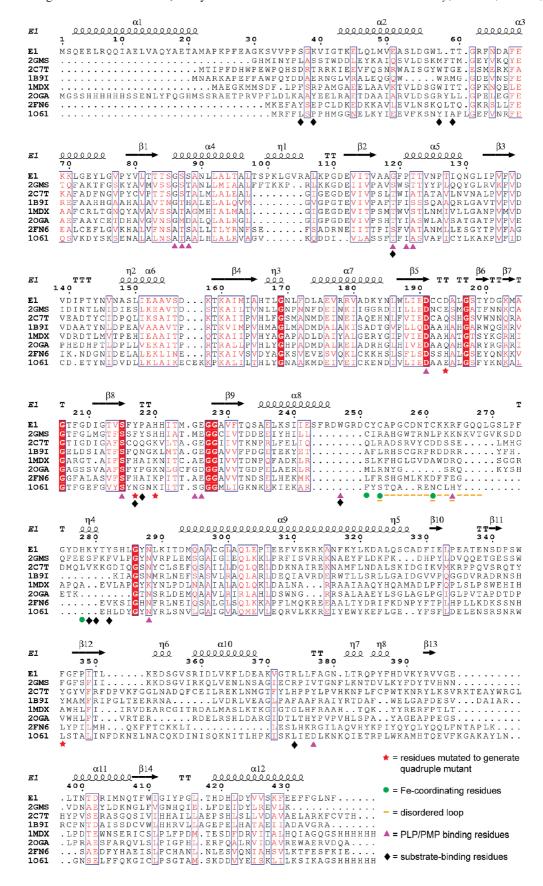


FIGURE 5: Sequence alignment of E₁ with homologous enzymes. All homologous sequences, identified by their PDB IDs, are aminotransferases (2C7T, 1MDX, 2OGA, 2FN6, and 1O6I) or dehydrases (2GMS and 1B9I). Symbols are used to highlight notable structural features of E_1 discussed in the text, where red stars = residues mutated in order to observe aminotransferase activity, green circles = iron binding residues, pink triangles = PLP/PMP-binding residues, orange dashes = disordered loop residues, and black diamonds = substrate binding residues.

 α -helix D macrodipole also helps to stabilize the anionic cofactor phosphate. The highly conserved D191 forms a salt bridge with the pyridinium nitrogen of PLP stabilizing the protonated state of the cofactor, a key to the "electron sink" property of PLP/PMP (Figure 4C) (1, 2). It is intriguing that E_1 has such unique catalytic capacities considering that residues important for cofactor bindings are highly conserved between E_1 and homologous PLP/PMP enzymes (Figure 5).

Substrate Binding Pocket. Compared to the highly conserved PLP/PMP pocket in E₁, the putative sugar substrate binding site is less conserved. Despite repetitive efforts to collect diffraction data from E₁-PLP-sugar cocrystals, we could not obtain the full occupancy of the amino sugar (CDP-4,6-dideoxy-4-amino-D-glucose, 8). However, the $F_{\rm o}-F_{\rm c}$ omit map (contoured at 3.0σ) of the amino sugar-bound structure has well-defined density of PLP, as well as partial electron density of the CDP-sugar moiety. This positions the amino sugar in between the electron density of PLP and CDP, and the sugar location is highly consistent with docking simulation of the amino sugar 8 into the H220K mutant structure. In 20–30 independent rounds of sugar–E₁ docking simulations (with no bias), we consistently identify a docking mode that extends from the cofactor C4' nitrogen toward the enzyme surface (Figure 6D). This docking mode overlaps consistently with the above broken electron density observed in the H220K E₁-PLP-sugar cocrystal data (Figure 6A,B,E). On the basis of the docking results and partial electron density of the PLP-sugar, we can identify residues proximal to the sugar substrate. Sequence alignment with the homologous enzymes revealed that these sugar binding residues are only partially conserved (Figures 5 and 6F). Specifically, the ribose-interacting residues K39-Y217-P218 of monomer A are not conserved, and the diphosphate-interacting residues such as S37-R374 of monomer A and K279-Y280 of monomer B are semiconserved, especially K279-Y280. Because E_1 is the only CDP-sugar binding enzyme among the homologous structures, the cytidine-interacting residues such as K39 of monomer A and W57-T59 of monomer B are the least conserved residues in the pocket. That many of the sugar recognition residues are not conserved is not surprising, because these variations are necessary to permit different substrate specificities in different enzymes. Interestingly, the majority of residues interacting with the sugar moiety are semiconserved aromatic residues, including F120-Y217 of monomer A and W247-Y280-Y282 of monomer B that interact with the sugar backbone and hydroxyl groups via hydrogen bonds (Figure 5).

Fe-S-Coordinating Loop. E₁ exhibits a unique [2Fe-2S] coordination motif that currently has no structural precedent. These coordinating residues, identified by systematic mutations of the E₁ cysteines, indicated that C251, C253, and C261 are essential for retaining enzyme activity and preserving the [2Fe-2S] cluster (15). The residues lie on a disordered loop in both the wild-type and H220K E₁ crystal structures (residues 253–267, Figure 3) located at the protein surface. While no [2Fe-2S] cluster is visible in the crystal structures, the location of one iron of the iron–sulfur center can be identified from a large peak in the anomalous difference electron density map calculated from the MAD data (Fe peak wavelength at 1.74 Å, Table S1, Figure 7A), with a signal-to-noise ratio of 5.6. The anomalous iron peak lies near the 253–268 loop region.

Previous mutagenesis showed that double alanine mutations of C192 and C193 resulted in the loss of iron-binding capability, suggesting that the fourth Fe-coordinating residue is C192 or C193, and C192/C193 can substitute for one another. The positions of C192 and C193 are well-defined in the apo and H220K E1 structures. However, to our surprise, these two cysteines are buried at the bottom of the E_1 active site, inaccessible to solvent or the would-be surface [2Fe-2S] cluster. Hence, neither C192 nor C193 likely plays a role in Fe coordination. A highly improbable large rearrangement of the protein structure is required (22 Å, Figure 7C) to move C192/C193 close to the other iron ligands (C251/C253/C261). The loss of iron binding in the C192A and C193A mutants is probably a result of mutation-induced conformational change, rather than a direct loss of the ironcoordinating cysteine. Interestingly, the crystal structure of E₁ reveals that the fourth ligand is likely H278, which occupies an iron coordination site, along with the proposed C251 and C253 (Figure 7A,B) (15). An omit map $(F_o - F_c)$ contoured at 3.0σ) displays clear density for the H278 side chain, and additional, less well defined density stretches from the proposed iron site indicate protein residues (albeit in disorder) in this region (Figure 7A,B). To examine the role of H278, the H278A and H278C mutants were prepared. As expected, the H278A E₁ mutant is incapable of holding iron and is catalytically inactive. In contrast, the H278C E₁ mutant retains approximately 50% of the iron content and the catalytic activity when compared to the wild-type E_1 . These results further substantiate the assigned role of H278 in iron-sulfur cluster binding.

The disorder in the 253-267 loop region of E_1 is unusual among AAT enzyme structures. Except ArnB (25), which has an 11-residue gap, all other homologous enzyme structures have ordered loops in this region, including ColD (22), BtrR (23), DesV (26), PseC (27), and AHBA synthase (13). This loop region is the least conserved region between E_1 and the other six enzymes (Figure 5). When the structure of E₁ is superimposed on the homologous structures (Figure 7D), it becomes obvious that the 253-267 loop region of one monomer, if connected, is adjacent to the active site of the other monomer. The well-studied aspartate transaminase is active only in the open conformation (2). Hence, a possible explanation for the loop flexibility observed in E₁ is that it allows "open" and "closed" conformational changes of the substrate binding pocket as a gating mechanism. However, in the homologous ColD (22), BtrR (23), and especially PseC (27), whose apo and substrate/inhibitor-bound structures are available for comparisons, this loop region remains unchanged with and without substrate binding. These observations suggest that a second possible reason for the E_1 loop flexibility is to allow close contact with the reductase E₃ to form a reasonably tight binary complex (29). Such a complex would facilitate electron transfer from the reduced iron-sulfur center of E₃ via the iron-sulfur cluster of E₁ to reduce the dehydration product $\mathbf{4}$ in the E_1 active site.

Structural Basis for the Dehydrase and Aminotransferase Activities. While the wild-type E_1 is a dehydrase, many of its close homologues are aminotransferases, such as ColD (22), BtrR (23), ArnB (25), DesV (26), and PseC (27). The two most obvious distinctions between E_1 and the AAT relatives are the presence of the [2Fe-2S] cluster and the switch of the active site lysine to a histidine (H220) in E_1

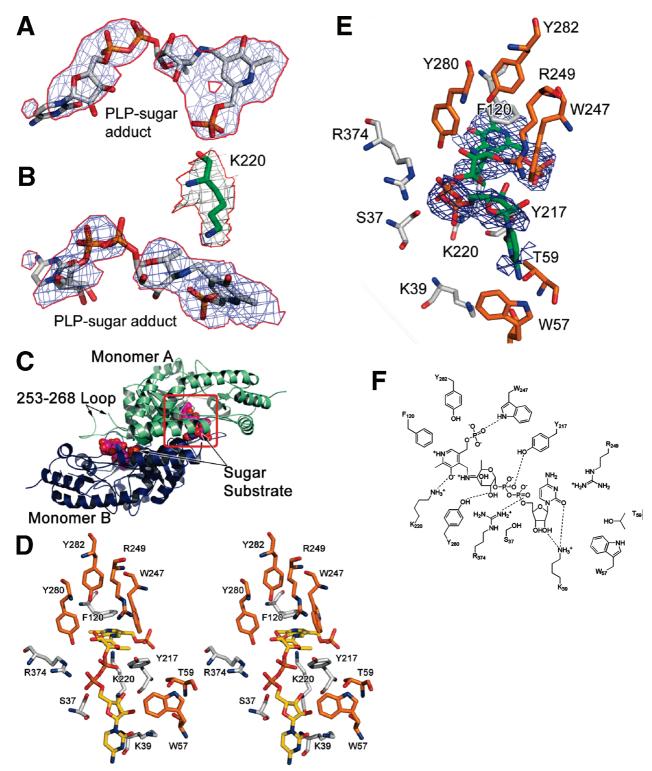


FIGURE 6: (A, B) Two views of the amino sugar-bound structure (CDP-4,6-dideoxy-4-amino-D-glucose, 8) omit map ($F_0 - F_c$, contoured at 3.0 σ), showing partially defined density of CDP-sugar and well-defined density of PLP. Note that the density between the sugar and CDP is broken, but the density of PLP and CDP orients the sugar moiety to the position shown in (A) and (B), which is also consistent with docking simulation of the sugar-CDP into E_1 . (C) The overall view of the docked sugar substrate 8 in E_1 dimer shows that the bound sugar extends from the coenzyme binding pocket toward the protein surface. (D) Stereo image of the docked full substrate (CDP-4,6-dideoxy-4-amino-D-glucose, 8). (E) Similar perspective as in (D), with the omit map density around the external aldimine. (F) Scheme of protein contacts with the substrate based upon the docking result.

(15). Prior to solving the E₁ crystal structures, in a retroevolutionary attempt to convert E1 from a dehydrase to an aminotransferase, we constructed the E₁ H220K mutant and found that H220K E₁ acts as a PLP-dependent transaminase with no dehydrase activity (17). We reasoned that the newly introduced K220 could form an external aldimine (9) with

an incoming amino sugar substrate (8) (Figure 2). The reaction proceeds via tautomerization to give 3, which after hydrolysis produces PMP and a keto sugar product (1). While this single mutation altered the function of E1, the reaction is unfortunately not catalytic since PLP cannot be regenerated after each catalytic cycle (Figure 2, path A). With the E₁

FIGURE 7: (A) The anomalous iron peak coordinate (green sphere) and electron density calculated with the MAD data (purple, contoured at 5 sigma) lie near the 253-268 loop region. (A, B) The omit map in gray density and highlighted in green for the H278, C251, and C253 side chains ($F_0 - F_c$ omit map contoured at 3σ) displays clear density of H278, and additional, less well defined density stretching from the proposed iron site indicates protein residues (albeit in disorder) in this region (outlined in red). (B) is an overhead perspective relative to (A), where the difference in the residues displayed in each view is for the sake of clarity. (C) The distance of C192 and C193 is 22-30 Å away from the main [Fe-S] coordinating loop. Hence, the fourth ligand is more likely H278. (D) The E₁ dimer with superimposition of loops from homologues with order in this region [2FN6 (blue), 2GMS (green)] and the disordered E₁ loop in red.

crystal structures in hand, we can dissect the molecular features that discriminate between dehydrase and aminotransferase activities.

Among the above homologous aminotransferases, ArnB (25) works on a uridine 5'-diphosphate ketopyranose (PDB codes 1MDX, 1MDO, 1MDZ, 1O61, and 1O62), which closely resembles the substrate for E_1 (Figure 8A). For this reason, we chose ArnB as the aminotransferase model for comparison with E_1 . Four putative active site residues were identified that are different between E_1 and ArnB: D194, Y217, H220, and F345 in E_1 and the corresponding H163, H185, K188, and H297 in ArnB (Figure 5).

To examine if these four residues are important for the aminotransferase activity of ArnB and if E₁ can be imparted this catalytic capacity, we generated the D194H-Y217H-H220K-F345H quadruple mutant and incubated a catalytic amount of the mutant enzyme with a large excess of PLP and L-glutamate in the presence of the biological substrate, CDP-4-keto-6-deoxy-D-glucose (1; see Figure 8B for the HPLC trace). HPLC analysis of the incubation mixture revealed the formation of a new product which was identified as CDP-4,6-dideoxy-4-amino-D-galactose (12), based on its retention time, which is identical to that of a standard (17), and high-resolution ESI-MS data (calcd for

 $C_{15}H_{25}N_4O_{14}P_2[M\cdot H]^-$, 547.0843; found, 547.0841). The control experiment conducted with E_1 -plasmid-free competent cells did not produce any turnover product. Thus, the quadruple mutation has successfully converted E_1 into a catalytic, PLP/L-glutamate-dependent transaminase that closely mimics the ArnB activity.

E₁ Quadruple Mutant Kinetic Efficiency. Upon determining that the E₁ quadruple mutant could produce the amino sugar 12, we sought to compare its kinetic efficiency relative to the ArnB aminotransferase. The two half-reactions are monitored using the distinct absorbance maxima of enzyme— PMP (E-PMP) and enzyme-PLP (E-PLP) at 340 and 430 nm, respectively (Figure 9). The first half-reaction involves amino transfer from L-glutamic acid to the E-PLP complex to generate E-PMP and α -ketoglutarate. In the second halfreaction, adding the CDP-keto sugar substrate 1 to the E-PMP results in amino transfer from the E-PMP to generate the amino sugar 12 and regenerate E-PLP (Figure 9). In the first half-raction, the curves fitted to the data in Figure 9 give consistently observed rate constants of 0.0266 \pm 0.002 and 0.0277 \pm 0.0021 min⁻¹ for data obtained at 340 and 430 nm, respectively. The observed rate constants published for ArnB aminotransferase in the first half-reaction were $0.065 \pm 0.014 \text{ min}^{-1}$ and $0.051 \pm 0.013 \text{ min}^{-1}$ at 340

α-KG

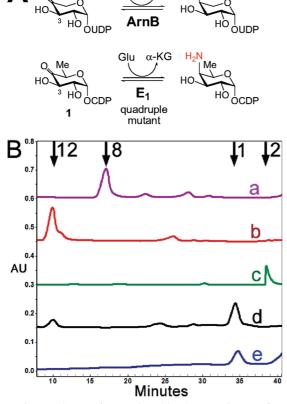


FIGURE 8: (A) The reaction catalyzed by ArnB aminotransferase is essentially the same as the E₁ quadruple mutant. (B) HPLC trace of the reaction mixture after incubation of the E_1 quadruple mutant with L-glutamate and substrate 1. Traces from the top: (a) the sugar standards 8 and (b) 12, followed by the (c) E_1-E_3 product 2, the incubation mixture of the quadruple mutant (d), and the control empty plasmid cell extract (e). The quadruple mutant distinctly produced the CDP-4,6-dideoxy-4-aminogalactose 12.

and 430 nm, respectively (25), in which the amino donor is L-methionine. Therefore, the reported activity of ArnB aminotransferase is comparable in magnitude to the observe activity of the E_1 quadruple mutant for the first half-reaction.

The second half activity of ArnB is not reported in the literature, although Breazeale et al. reported the overall ArnB activity as 1.3×10^3 nmol min⁻¹ mg⁻¹ for the appearance of the amino sugar product (30). Since the second half rate constant in the E₁ quadruple mutant is 10 times slower than the first half rate constant, it appears that, for the E₁ quadruple mutant, the rate-limiting step is in the second half-reaction, which approximates the overall turnover. In the second halfreaction, the E₁ quadruple mutant has a $k_{\rm obs}$ of 0.0035 \pm 0.0001 min⁻¹, which can be converted to 9.6 nmol min⁻¹ mg⁻¹ for the appearance of E-PLP, assuming the E-PLP extinction coefficient is 8050 at A_{430} , as reported in the literature (31). Therefore, the overall rates of ArnB and the E₁ quadruple mutant are differed by 2 orders of magnitude, which may be caused by the nature of the protein mutations or by the different binding ability of the substrate/transition state because ArnB and E₁ have different sugar substrates (Figure 8A), and the sugar binding residues in ArnB and E₁ are not conserved (Figure 5).

At the Crossroads of Transamination, Dehydration, and *Epimerization.* The above result is distinctly different from what was observed with the single H220K mutation that allowed noncatalytic conversion of CDP-4,6-dideoxy-4-

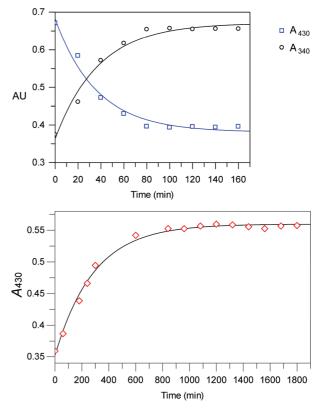


FIGURE 9: Spectrophotometric assays of the E₁ quadruple mutant to monitor the first and second half-reactions of stepwise amino transfer from L-glutamate \rightarrow E-PLP \rightarrow substrate sugar 1. The top panel displays the plot monitoring the first half-reaction via the disappearance of E-PLP (A_{430}) and the concurrent appearance of E-PMP (A_{340}) over time, as L-glutamic acid transfers its amide nitrogen to E-PLP to generate E-PMP and α-ketoglutarate (individual data points represent the average of three collected at each time point). The bottom panel displays the plot following the second half-reaction via appearance of E-PLP (A_{430}) in the presence of E-PMP and substrate 1, where the amino group from PMP is transferred to the substrate to yield product 12 and to regenerate E-PLP, thus finishing a catalytic cycle (individual data points represent the average of three collected at each time point). Observed rate constants are tabulated in Table 2.

amino-D-glucose (8), but not CDP-4,6-dideoxy-4-amino-D-galactose (12), into 1 (17). Figure 2 summarizes the chemistry catalyzed by E₁ and its mutants. The wild-type E_1 catalyzed reaction follows the route $1 \rightarrow 3 \rightarrow 10 \rightarrow 4 \rightarrow$ $6 \rightarrow 2$ (Figure 2, path C), starting with the formation of a Schiff base 3 between PMP and the substrate 1, followed by C4' deprotonation to form the quinonoid intermediate 10, β -elimination of the 3-OH group to give 4, and then reduction in the presence of reductase E₃ to generate 2 via 6. The overall conversion is a PMP-dependent C-3 deoxygenation $(1 \rightarrow 2)$. The reaction mediated by H220K E₁ proceeds via the route $8 \rightarrow 9 \rightarrow 10 \rightarrow 3 \rightarrow 1$ (Figure 2, path A), starting with the exchange of the internal aldimine 7 formed between PLP and K220 to the external aldimine 9 formed between PLP and the amino sugar 8. Subsequently, deprotonation at C4 of the sugar substrate generates the quinonoid intermediate 10. The final steps involve the reprotonation at C4' of the cofactor to generate 3, which is then hydrolyzed to produce 1. The overall transformation is a transamination reaction ($8 \rightarrow 1$). However, the reaction is not catalytic since the coenzyme is not regenerated. In contrast, the quadruple mutant-catalyzed reaction $(1 \rightarrow 12)$ is fully catalytic and is initiated by the transamination of PLP and L-glutamate to

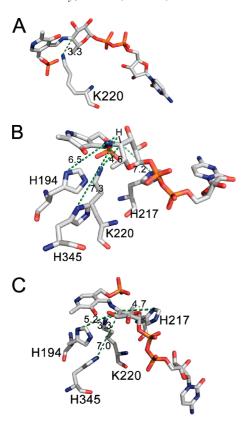


FIGURE 10: Docking simulations of (A) the PLP-CDP-6-deoxy-4-aminoglucose adduct **9** that shows K220 is in close proximity to the C4 axial position of the sugar substrate, (B) the PLP-CDP-6-deoxy-4-amiongalactose adduct **11** in the quadruple mutant, in which H194 and H194, instead of K220, are the likely acid/base residues, and (C) the quinonoid intermediate **10** in the quadruple mutant revealing that both H217 and H194 promote equatorial protonation at C4 of **10** to form **11**. Distances are in angstroms.

produce PMP and α -ketoglutarate. The subsequent steps, following the route $1 \rightarrow 3 \rightarrow 10 \rightarrow 11 \rightarrow 12$ (Figure 1, path B), include formation of the Schiff base 3 between PMP and 1, deprotonation at C4' of the cofactor to form the quinonoid intermediate 10, reprotonation at C4 of the sugar substrate from the equatorial position to give the aldimine 11, and transamination with K220 to give the product 12 and the internal aldimine 7. It is evident that the quinonoid intermediate 10 sits at the crossroads of transamination, epimerization, tautomerization, and β -elimination, and the outcomes of the reaction depend on the regioselectivity of the protonation/deprotonation processes mediated by different active site residues

To identify residues responsible for protonation/deprotonation in the preceding three reaction types, we carried out extensive *in silico* docking of the coenzyme—sugar adducts **9**, **10**, and **11** into the active sites of the H220K E_1 and the quadruple mutant homology model (Figure 10). Our results show that each coenzyme—substrate adduct has a unique docking mode and can help to explain the different chemical outcomes. Docking **9** or **10**, the best result places K220 close to the hexose C4 axial position (Figure 10A), allowing for pathway $\mathbf{8} \rightarrow \mathbf{9} \rightarrow \mathbf{10} \rightarrow \mathbf{3} \rightarrow \mathbf{1}$, with K220 serving as the active base/acid for tautomerization. The result for adduct **11**, with either H220K E_1 or the quadruple mutant, yielded a specific binding mode positioning K220 away from the C4 axial position (Figure 10B). Docking of **10** into the quadruple mutant shows that

either H217 or H194 can protonate substrate **10** at the C4 equatorial position to generate **11** (Figure 10C), leading to product **12**. H220K E₁ cannot carry out this reaction lacking these two histidines. None of the above residues exist in the active site of the wild-type E₁ to carry out C4 protonation, where dehydration (via C3-OH protonation) takes place instead, presumably activated by a water, hydrogen bonded to Y280 or Y282.

Biological Significance. The wild-type and H220K-PLP E₁ crystal structures provide new insights into the architecture and chemistry employed by this dehydrase. The three-dimensional picture allows visualization of the active site, which corroborates the role of H220 as a catalytic acid/base, and identifies residues important for transamination/dehydration/epimerization. One of the unusual features of the E_1 dehydrase is the [2Fe-2S] cluster. The crystal structures have located the Fe site for the first time on the protein surface, as well as clarified residues that coordinate to the Fe. When compared with homologous structures, the E1 structures also enable us to mutagenically convert the enzyme into a catalytic, PLP/ L-glutamate-dependent aminotransferase. The significance of the quadruple mutant results are twofold: the mutant identifies the four key active site residues necessary to distinguish between a dehydrase and an aminotransferase, and, for the first time, it is possible to correlate mutated residues in the two different E₁ mutants (the quadruple versus H220K) with different specificities toward epimeric sugars (CDP-4,6-dideoxy-4-aminoglucose versus CDP-4,6dideoxy-4-aminogalactose). In terms of medical applications, E₁ is responsible for making the progenitor to at least four of the seven naturally occurring 3,6-dideoxyhexoses, which in turn confer various pathogenic capabilities to harboring Gram-negative bacteria. The E1 crystal structures reported herein provide a foundation for the design of inhibitors for the biosynthesis of 3,6-dideoxyhexoses in Gram-negative bacteria.

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SUPPORTING INFORMATION AVAILABLE

Table S1 organizing the MAD crystallographic data used to solve the iron site along with refinement statistics. This material is available free of charge via the Internet at http://pubs.acs.org.

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